

# Isophthalic acid, isoheptyl 2-isopropoxyphenyl ester

Inchi:	InChI=1S/C23H28O5/c1-16(2)9-8-14-26-22(24)18-10-7-11-19(15-18)23(25)28-21-13-6-5
InchiKey:	KFLONDAZBWJKAY-UHFFFAOYSA-N
Formula:	C23H28O5
SMILES:	CC(C)CCCOC(=O)c1cccc(C(=O)Oc2ccccc2OC(C)C)c1
Mol. weight [g/mol]:	384.47

## Physical Properties

Property code	Value	Unit	Source
gf	-229.38	kJ/mol	Joback Method
hf	-700.31	kJ/mol	Joback Method
hfus	42.35	kJ/mol	Joback Method
hvap	92.61	kJ/mol	Joback Method
log10ws	-6.72		Crippen Method
logp	5.286		Crippen Method
mvol	308.160	ml/mol	McGowan Method
pc	1368.70	kPa	Joback Method
rinpol	2838.00		NIST Webbook
rinpol	2838.00		NIST Webbook
tb	963.08	K	Joback Method
tc	1190.21	K	Joback Method
tf	563.40	K	Joback Method
vc	1.161	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	984.26	J/molxK	963.08	Joback Method
cpg	997.72	J/molxK	1000.94	Joback Method
cpg	1009.59	J/molxK	1038.79	Joback Method
cpg	1019.92	J/molxK	1076.65	Joback Method
cpg	1028.71	J/molxK	1114.50	Joback Method
cpg	1036.01	J/molxK	1152.36	Joback Method
cpg	1041.83	J/molxK	1190.21	Joback Method
dvisc	0.0002788	Paxs	563.40	Joback Method

dvisc	0.0001480	Paxs	630.01	Joback Method
dvisc	0.0000887	Paxs	696.63	Joback Method
dvisc	0.0000581	Paxs	763.24	Joback Method
dvisc	0.0000407	Paxs	829.85	Joback Method
dvisc	0.0000301	Paxs	896.47	Joback Method
dvisc	0.0000232	Paxs	963.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344430&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344430&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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