

Phthalic acid, heptyl 2-phenoxyethyl ester

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|-----------------------------|--|
| Inchi: | InChI=1S/C23H28O5/c1-2-3-4-5-11-16-27-22(24)20-14-9-10-15-21(20)23(25)28-18-17-2 |
| InchiKey: | AWRVSFQJPTXQLT-UHFFFAOYSA-N |
| Formula: | C23H28O5 |
| SMILES: | CCCCCCCOC(=O)c1ccccc1C(=O)OCCOc1ccccc1 |
| Mol. weight [g/mol]: | 384.47 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -214.87 | kJ/mol | Joback Method |
| hf | -678.28 | kJ/mol | Joback Method |
| hfus | 49.78 | kJ/mol | Joback Method |
| hvap | 92.73 | kJ/mol | Joback Method |
| log10ws | -6.24 | | Crippen Method |
| logp | 5.050 | | Crippen Method |
| mvol | 308.160 | ml/mol | McGowan Method |
| pc | 1367.69 | kPa | Joback Method |
| rinpol | 3282.00 | | NIST Webbook |
| rinpol | 3282.00 | | NIST Webbook |
| tb | 958.98 | K | Joback Method |
| tc | 1182.01 | K | Joback Method |
| tf | 580.88 | K | Joback Method |
| vc | 1.173 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 984.53 | J/molxK | 958.98 | Joback Method |
| cpg | 1037.67 | J/molxK | 1144.84 | Joback Method |
| cpg | 1029.87 | J/molxK | 1107.66 | Joback Method |
| cpg | 1020.69 | J/molxK | 1070.49 | Joback Method |
| cpg | 1010.09 | J/molxK | 1033.32 | Joback Method |
| cpg | 998.05 | J/molxK | 996.15 | Joback Method |
| cpg | 1044.12 | J/molxK | 1182.01 | Joback Method |
| dvisc | 0.0000273 | Paxs | 958.98 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000349 | Paxs | 895.96 | Joback Method |
| dvisc | 0.0000464 | Paxs | 832.95 | Joback Method |
| dvisc | 0.0000644 | Paxs | 769.93 | Joback Method |
| dvisc | 0.0000950 | Paxs | 706.91 | Joback Method |
| dvisc | 0.0001511 | Paxs | 643.90 | Joback Method |
| dvisc | 0.0002656 | Paxs | 580.88 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U382483&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|----------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log₁₀ws: | Log ₁₀ of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m_{cvol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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