

# Heptanoic acid, 2-acetyl-, ethyl ester

<b>Other names:</b>	Ethyl 2-acetylheptanoate
<b>Inchi:</b>	InChI=1S/C11H20O3/c1-4-6-7-8-10(9(3)12)11(13)14-5-2/h10H,4-8H2,1-3H3
<b>InchiKey:</b>	XIGZBCUFFUBWDM-UHFFFAOYSA-N
<b>Formula:</b>	C11H20O3
<b>SMILES:</b>	<chem>CCCCC(C(C)=O)C(=O)OCC</chem>
<b>Mol. weight [g/mol]:</b>	200.27
<b>CAS:</b>	24317-94-0

## Physical Properties

Property code	Value	Unit	Source
gf	-323.54	kJ/mol	Joback Method
hf	-633.03	kJ/mol	Joback Method
hfus	25.11	kJ/mol	Joback Method
hvap	55.59	kJ/mol	Joback Method
log10ws	-2.33		Crippen Method
logp	2.335		Crippen Method
mcvol	174.860	ml/mol	McGowan Method
pc	2149.31	kPa	Joback Method
tb	580.80	K	Joback Method
tc	763.33	K	Joback Method
tf	320.82	K	Joback Method
vc	0.675	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	439.02	J/molxK	580.80	Joback Method
cpg	453.42	J/molxK	611.22	Joback Method
cpg	467.17	J/molxK	641.64	Joback Method
cpg	480.30	J/molxK	672.07	Joback Method
cpg	492.80	J/molxK	702.49	Joback Method
cpg	504.69	J/molxK	732.91	Joback Method
cpg	515.97	J/molxK	763.33	Joback Method
dvisc	0.0032667	Paxs	320.82	Joback Method

dvisc	0.0015567	Paxs	364.15	Joback Method
dvisc	0.0008685	Paxs	407.48	Joback Method
dvisc	0.0005420	Paxs	450.81	Joback Method
dvisc	0.0003675	Paxs	494.14	Joback Method
dvisc	0.0002652	Paxs	537.47	Joback Method
dvisc	0.0002010	Paxs	580.80	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24317940&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24317940&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/88-648-6/Heptanoic-acid-2-acetyl-ethyl-ester.pdf>

Generated by Cheméo on 2024-04-26 04:44:16.308956811 +0000 UTC m=+16395905.229534127.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.