

# 4-Methoxy-6-((1S,3aR,4S,6aR)-4-(3,4,5-trimethoxy

<b>Inchi:</b>	InChI=1S/C23H26O8/c1-24-16-5-12(6-17(25-2)22(16)27-4)20-14-9-29-21(15(14)10-28-2
<b>InchiKey:</b>	DHWUVPPRBIJJKS-UHFFFAOYSA-N
<b>Formula:</b>	C23H26O8
<b>SMILES:</b>	COc1cc(C2OCC3C(c4cc(OC)c5c(c4)OCO5)OCC23)cc(OC)c1OC
<b>Mol. weight [g/mol]:</b>	430.45
<b>CAS:</b>	77394-27-5

## Physical Properties

Property code	Value	Unit	Source
gf	-304.32	kJ/mol	Joback Method
hf	-984.95	kJ/mol	Joback Method
hfus	69.02	kJ/mol	Joback Method
hvap	102.77	kJ/mol	Joback Method
log10ws	-4.65		Crippen Method
logp	3.525		Crippen Method
mcvol	301.790	ml/mol	McGowan Method
pc	1536.66	kPa	Joback Method
rinpol	3431.00		NIST Webbook
rinpol	3431.00		NIST Webbook
tb	1030.45	K	Joback Method
tc	1275.27	K	Joback Method
tf	714.67	K	Joback Method
vc	1.117	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1063.08	J/molxK	1030.45	Joback Method
cpg	1076.20	J/molxK	1071.25	Joback Method
cpg	1087.61	J/molxK	1112.06	Joback Method
cpg	1097.35	J/molxK	1152.86	Joback Method
cpg	1105.48	J/molxK	1193.66	Joback Method
cpg	1112.04	J/molxK	1234.46	Joback Method
cpg	1117.08	J/molxK	1275.27	Joback Method

dvisc	0.0008210	Paxs	714.67	Joback Method
dvisc	0.0006607	Paxs	767.30	Joback Method
dvisc	0.0005468	Paxs	819.93	Joback Method
dvisc	0.0004629	Paxs	872.56	Joback Method
dvisc	0.0003994	Paxs	925.19	Joback Method
dvisc	0.0003502	Paxs	977.82	Joback Method
dvisc	0.0003111	Paxs	1030.45	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C77394275&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C77394275&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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