

# Pimelic acid, hexyl 2-naphthyl ester

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C23H30O4/c1-2-3-4-10-17-26-22(24)13-6-5-7-14-23(25)27-21-16-15-19-11-8-9 |
| InchiKey:            | NDEADIRDUBEQSX-UHFFFAOYSA-N                                                       |
| Formula:             | C23H30O4                                                                          |
| SMILES:              | CCCCCOC(=O)CCCCC(=O)Oc1ccc2ccccc2c1                                               |
| Mol. weight [g/mol]: | 370.48                                                                            |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -115.63 | kJ/mol  | Joback Method  |
| hf            | -591.52 | kJ/mol  | Joback Method  |
| hfus          | 51.57   | kJ/mol  | Joback Method  |
| hvap          | 89.68   | kJ/mol  | Joback Method  |
| log10ws       | -7.06   |         | Crippen Method |
| logp          | 5.819   |         | Crippen Method |
| mcvol         | 306.590 | ml/mol  | McGowan Method |
| pc            | 1300.47 | kPa     | Joback Method  |
| rinpol        | 2987.00 |         | NIST Webbook   |
| rinpol        | 2987.00 |         | NIST Webbook   |
| tb            | 928.86  | K       | Joback Method  |
| tc            | 1143.95 | K       | Joback Method  |
| tf            | 564.93  | K       | Joback Method  |
| vc            | 1.185   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 979.65    | J/mol×K | 928.86          | Joback Method |
| cpg           | 1044.64   | J/mol×K | 1108.10         | Joback Method |
| cpg           | 1033.68   | J/mol×K | 1072.25         | Joback Method |
| cpg           | 1021.76   | J/mol×K | 1036.41         | Joback Method |
| cpg           | 1008.82   | J/mol×K | 1000.56         | Joback Method |
| cpg           | 994.80    | J/mol×K | 964.71          | Joback Method |
| cpg           | 1054.71   | J/mol×K | 1143.95         | Joback Method |
| dvisc         | 0.0000729 | Pa×s    | 928.86          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0000906 | Pa×s | 868.20 | Joback Method |
| dvisc | 0.0001164 | Pa×s | 807.55 | Joback Method |
| dvisc | 0.0001558 | Pa×s | 746.89 | Joback Method |
| dvisc | 0.0002195 | Pa×s | 686.24 | Joback Method |
| dvisc | 0.0003305 | Pa×s | 625.59 | Joback Method |
| dvisc | 0.0005434 | Pa×s | 564.93 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U416712&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U416712&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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