

# 2-hydroxy-3,5-dimethoxy-9,10-dihydrophenanthrene

InChI: **acetylated** InChI=1S/C18H18O4/c1-11(19)22-17-9-13-8-7-12-5-4-6-15(20-2)18(12)14(13)10-16(17)2  
InChIKey: MPPPUKPVQACBDB-UHFFFAOYSA-N

Formula: C18H18O4

SMILES: COc1cc2c(cc1OC(C)=O)CCc1cccc(OC)c1-2

Mol. weight [g/mol]: 298.33

## Physical Properties

Property code	Value	Unit	Source
gf	-86.01	kJ/mol	Joback Method
hf	-409.08	kJ/mol	Joback Method
hfus	32.84	kJ/mol	Joback Method
hvap	77.55	kJ/mol	Joback Method
log10ws	-5.29		Crippen Method
logp	3.395		Crippen Method
mcvol	225.280	ml/mol	McGowan Method
pc	2049.31	kPa	Joback Method
rinpol	2375.00		NIST Webbook
rinpol	2375.00		NIST Webbook
tb	817.77	K	Joback Method
tc	1047.83	K	Joback Method
tf	550.38	K	Joback Method
vc	0.854	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	652.62	J/molxK	817.77	Joback Method
cpg	666.95	J/molxK	856.11	Joback Method
cpg	680.21	J/molxK	894.46	Joback Method
cpg	692.44	J/molxK	932.80	Joback Method
cpg	703.69	J/molxK	971.14	Joback Method
cpg	713.99	J/molxK	1009.49	Joback Method
cpg	723.39	J/molxK	1047.83	Joback Method
dvisc	0.0006403	Paxs	550.38	Joback Method

dvisc	0.0004913	Paxs	594.95	Joback Method
dvisc	0.0003911	Paxs	639.51	Joback Method
dvisc	0.0003207	Paxs	684.08	Joback Method
dvisc	0.0002695	Paxs	728.64	Joback Method
dvisc	0.0002310	Paxs	773.21	Joback Method
dvisc	0.0002014	Paxs	817.77	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R273873&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R273873&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/88-854-7/2-hydroxy-3-5-dimethoxy-9-10-dihydrophenanthrene-acetylated.pdf>

Generated by Cheméo on 2024-04-20 08:41:50.048099429 +0000 UTC m=+15891758.968676750.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.