

# Glutaric acid, 1-cyclopentylethyl 2-methylpent-3-yl ester

<b>Inchi:</b>	InChI=1S/C18H32O4/c1-5-16(13(2)3)22-18(20)12-8-11-17(19)21-14(4)15-9-6-7-10-15/h1
<b>InchiKey:</b>	FQSAFYUOADMTLL-UHFFFAOYSA-N
<b>Formula:</b>	C18H32O4
<b>SMILES:</b>	CCC(OC(=O)CCCC(=O)OC(C)C1CCCC1)C(C)C
<b>Mol. weight [g/mol]:</b>	312.44

## Physical Properties

Property code	Value	Unit	Source
gf	-337.93	kJ/mol	Joback Method
hf	-859.81	kJ/mol	Joback Method
hfus	31.32	kJ/mol	Joback Method
hvap	73.07	kJ/mol	Joback Method
log10ws	-4.72		Crippen Method
logp	4.256		Crippen Method
mcvol	268.500	ml/mol	McGowan Method
pc	1423.99	kPa	Joback Method
rinpol	2000.00		NIST Webbook
rinpol	2000.00		NIST Webbook
tb	777.78	K	Joback Method
tc	975.46	K	Joback Method
tf	402.84	K	Joback Method
vc	1.014	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	843.47	J/molxK	777.78	Joback Method
cpg	862.32	J/molxK	810.73	Joback Method
cpg	879.96	J/molxK	843.67	Joback Method
cpg	896.42	J/molxK	876.62	Joback Method
cpg	911.71	J/molxK	909.57	Joback Method
cpg	925.86	J/molxK	942.51	Joback Method
cpg	938.91	J/molxK	975.46	Joback Method
dvisc	0.0022101	Paxs	402.84	Joback Method

dvisc	0.0008789	Paxs	465.33	Joback Method
dvisc	0.0004348	Paxs	527.82	Joback Method
dvisc	0.0002496	Paxs	590.31	Joback Method
dvisc	0.0001594	Paxs	652.80	Joback Method
dvisc	0.0001101	Paxs	715.29	Joback Method
dvisc	0.0000807	Paxs	777.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405459&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405459&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/88-908-7/Glutaric-acid-1-cyclopentylethyl-2-methylpent-3-yl-ester.pdf>

Generated by Cheméo on 2024-04-26 03:09:49.224527707 +0000 UTC m=+16390238.145105030.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.