

# Carbonic acid, hexadecyl methyl ester

<b>Other names:</b>	Methyl hexadecyl carbonate
<b>Inchi:</b>	InChI=1S/C18H36O3/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-21-18(19)20-2/h3-17H2
<b>InchiKey:</b>	WZMKUTXQIPEPTH-UHFFFAOYSA-N
<b>Formula:</b>	C18H36O3
<b>SMILES:</b>	CCCCCCCCCCCCCCCCOC(=O)OC
<b>Mol. weight [g/mol]:</b>	300.48

## Physical Properties

Property code	Value	Unit	Source
gf	-238.24	kJ/mol	Joback Method
hf	-791.87	kJ/mol	Joback Method
hfus	46.35	kJ/mol	Joback Method
hvap	67.23	kJ/mol	Joback Method
log10ws	-6.28		Crippen Method
logp	6.251		Crippen Method
mcvol	277.790	ml/mol	McGowan Method
pc	1175.24	kPa	Joback Method
tb	709.95	K	Joback Method
tc	879.78	K	Joback Method
tf	387.01	K	Joback Method
vc	1.085	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	821.97	J/molxK	709.95	Joback Method
cpg	906.62	J/molxK	851.47	Joback Method
cpg	891.34	J/molxK	823.17	Joback Method
cpg	875.24	J/molxK	794.86	Joback Method
cpg	858.32	J/molxK	766.56	Joback Method
cpg	840.57	J/molxK	738.25	Joback Method
cpg	921.11	J/molxK	879.78	Joback Method
dvisc	0.0000729	Paxs	709.95	Joback Method
dvisc	0.0000973	Paxs	656.13	Joback Method

dvisc	0.0001367	Paxs	602.30	Joback Method
dvisc	0.0002052	Paxs	548.48	Joback Method
dvisc	0.0003367	Paxs	494.66	Joback Method
dvisc	0.0006234	Paxs	440.83	Joback Method
dvisc	0.0013700	Paxs	387.01	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U314627&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U314627&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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