

# 1,2,3-Propenetricarboxylic acid, tri(2-methylpropyl) ester

<b>Other names:</b>	Triisobutyl (1z)-1-propene-1,2,3-tricarboxylate
<b>Inchi:</b>	InChI=1S/C18H30O6/c1-12(2)9-22-16(19)7-15(18(21)24-11-14(5)6)8-17(20)23-10-13(3)4
<b>InchiKey:</b>	ZZDFZNPBXGCBBN-CHHVJCJISA-N
<b>Formula:</b>	C18H30O6
<b>SMILES:</b>	CC(C)COC(=O)C=C(CC(=O)OCC(C)C)C(=O)OCC(C)C
<b>Mol. weight [g/mol]:</b>	342.43
<b>CAS:</b>	10089-45-9

## Physical Properties

Property code	Value	Unit	Source
gf	-536.73	kJ/mol	Joback Method
hf	-1057.66	kJ/mol	Joback Method
hfus	39.06	kJ/mol	Joback Method
hvap	82.00	kJ/mol	Joback Method
log10ws	-3.07		Crippen Method
logp	2.901		Crippen Method
mcvol	282.500	ml/mol	McGowan Method
pc	1344.71	kPa	Joback Method
tb	842.83	K	Joback Method
tc	1040.87	K	Joback Method
tf	445.06	K	Joback Method
vc	1.079	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	876.99	J/molxK	842.83	Joback Method
cpg	892.55	J/molxK	875.84	Joback Method
cpg	907.00	J/molxK	908.84	Joback Method
cpg	920.36	J/molxK	941.85	Joback Method
cpg	932.64	J/molxK	974.86	Joback Method
cpg	943.86	J/molxK	1007.87	Joback Method
cpg	954.03	J/molxK	1040.87	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10089459&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10089459&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/89-101-1/1-2-3-Propenetricarboxylic-acid-tri-2-methylpropyl-ester.pdf>

Generated by Cheméo on 2024-04-17 02:22:31.196911341 +0000 UTC m=+15609800.117488656.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.