

# cis-Cyclohex-4-en-1,2-dicarboxylic acid, di(2-methylbutyl) ester

<b>Inchi:</b>	InChI=1S/C18H30O4/c1-5-13(3)11-21-17(19)15-9-7-8-10-16(15)18(20)22-12-14(4)6-2/h
<b>InchiKey:</b>	DZWIAYAUFVOENZ-UHFFFAOYSA-N
<b>Formula:</b>	C18H30O4
<b>SMILES:</b>	CCC(C)COC(=O)C1CC=CCC1C(=O)OCC(C)CC
<b>Mol. weight [g/mol]:</b>	310.43

## Physical Properties

Property code	Value	Unit	Source
gf	-325.34	kJ/mol	Joback Method
hf	-823.25	kJ/mol	Joback Method
hfus	35.03	kJ/mol	Joback Method
hvap	73.61	kJ/mol	Joback Method
log10ws	-3.86		Crippen Method
logp	3.747		Crippen Method
mvol	264.200	ml/mol	McGowan Method
pc	1445.73	kPa	Joback Method
rinpol	2019.00		NIST Webbook
rinpol	2019.00		NIST Webbook
tb	776.98	K	Joback Method
tc	977.20	K	Joback Method
tf	410.84	K	Joback Method
vc	0.998	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	823.15	J/molxK	776.98	Joback Method
cpg	841.87	J/molxK	810.35	Joback Method
cpg	859.31	J/molxK	843.72	Joback Method
cpg	875.50	J/molxK	877.09	Joback Method
cpg	890.44	J/molxK	910.46	Joback Method
cpg	904.16	J/molxK	943.83	Joback Method
cpg	916.66	J/molxK	977.20	Joback Method
dvisc	0.0016490	Paxs	410.84	Joback Method

dvisc	0.0007278	Paxs	471.86	Joback Method
dvisc	0.0003874	Paxs	532.89	Joback Method
dvisc	0.0002347	Paxs	593.91	Joback Method
dvisc	0.0001561	Paxs	654.93	Joback Method
dvisc	0.0001113	Paxs	715.96	Joback Method
dvisc	0.0000837	Paxs	776.98	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382697&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382697&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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