

# Benzenepropanoic acid, 3-phenylpropyl ester

<b>Other names:</b>	3-Phenylpropionic acid, 3-phenylpropyl ester 3-Phenylpropyl 3-phenylpropanoate 3-phenylpropyl 3-phenylpropionate
<b>Inchi:</b>	InChI=1S/C18H20O2/c19-18(14-13-17-10-5-2-6-11-17)20-15-7-12-16-8-3-1-4-9-16/h1-6,
<b>InchiKey:</b>	KLDUFQGXXVNTDL-UHFFFAOYSA-N
<b>Formula:</b>	C18H20O2
<b>SMILES:</b>	O=C(CCc1ccccc1)OCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	268.35
<b>CAS:</b>	60045-27-4

## Physical Properties

Property code	Value	Unit	Source
gf	91.58	kJ/mol	Joback Method
hf	-186.59	kJ/mol	Joback Method
hfus	33.24	kJ/mol	Joback Method
hvap	69.37	kJ/mol	Joback Method
log10ws	-4.43		Crippen Method
logp	3.795		Crippen Method
mcvol	224.400	ml/mol	McGowan Method
pc	1996.55	kPa	Joback Method
rinpol	2090.00		NIST Webbook
rinpol	2115.00		NIST Webbook
rinpol	2175.20		NIST Webbook
rinpol	2090.00		NIST Webbook
rinpol	2175.20		NIST Webbook
ripol	2925.00		NIST Webbook
ripol	2925.00		NIST Webbook
tb	740.89	K	Joback Method
tc	965.64	K	Joback Method
tf	417.62	K	Joback Method
vc	0.852	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	626.67	J/molxK	740.89	Joback Method
cpg	643.33	J/molxK	778.35	Joback Method
cpg	658.75	J/molxK	815.81	Joback Method
cpg	672.98	J/molxK	853.26	Joback Method
cpg	686.08	J/molxK	890.72	Joback Method
cpg	698.12	J/molxK	928.18	Joback Method
cpg	709.15	J/molxK	965.64	Joback Method
dvisc	0.0012396	Paxs	417.62	Joback Method
dvisc	0.0006357	Paxs	471.50	Joback Method
dvisc	0.0003739	Paxs	525.38	Joback Method
dvisc	0.0002427	Paxs	579.25	Joback Method
dvisc	0.0001696	Paxs	633.13	Joback Method
dvisc	0.0001253	Paxs	687.01	Joback Method
dvisc	0.0000968	Paxs	740.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C60045274&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C60045274&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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