

# decyl hydrogen phthalate

<b>Other names:</b>	2-(Decyloxycarbonyl)benzoic acid
<b>Inchi:</b>	InChI=1S/C18H26O4/c1-2-3-4-5-6-7-8-11-14-22-18(21)16-13-10-9-12-15(16)17(19)20/h
<b>InchiKey:</b>	FEFCILUKYGHITK-UHFFFAOYSA-N
<b>Formula:</b>	C18H26O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)c1ccccc1C(=O)O
<b>Mol. weight [g/mol]:</b>	306.40
<b>CAS:</b>	24539-60-4

## Physical Properties

Property code	Value	Unit	Source
gf	-296.20	kJ/mol	Joback Method
hf	-699.40	kJ/mol	Joback Method
hfus	44.50	kJ/mol	Joback Method
hvap	91.18	kJ/mol	Joback Method
log10ws	-5.54		Crippen Method
logp	4.682		Crippen Method
mcvol	255.600	ml/mol	McGowan Method
pc	1694.90	kPa	Joback Method
rinpol	2431.00		NIST Webbook
rinpol	2431.00		NIST Webbook
tb	865.24	K	Joback Method
tc	1065.69	K	Joback Method
tf	514.47	K	Joback Method
vc	0.985	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	792.77	J/molxK	865.24	Joback Method
cpg	851.18	J/molxK	1032.28	Joback Method
cpg	841.26	J/molxK	998.87	Joback Method
cpg	830.49	J/molxK	965.46	Joback Method
cpg	818.84	J/molxK	932.06	Joback Method
cpg	806.28	J/molxK	898.65	Joback Method

cpg	860.28	J/mol×K	1065.69	Joback Method
dvisc	0.0000148	Paxs	865.24	Joback Method
dvisc	0.0000213	Paxs	806.78	Joback Method
dvisc	0.0000323	Paxs	748.32	Joback Method
dvisc	0.0000528	Paxs	689.86	Joback Method
dvisc	0.0000942	Paxs	631.39	Joback Method
dvisc	0.0001895	Paxs	572.93	Joback Method
dvisc	0.0004465	Paxs	514.47	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24539604&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24539604&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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