

# Butyric acid, 2-phenyl-, oct-3-en-2-yl ester

<b>Inchi:</b>	InChI=1S/C18H26O2/c1-4-6-7-9-12-15(3)20-18(19)17(5-2)16-13-10-8-11-14-16/h8-15,17
<b>InchiKey:</b>	PRQHFUJYDIPSFY-FMIVXFBMSA-N
<b>Formula:</b>	C18H26O2
<b>SMILES:</b>	CCCCC=CC(C)OC(=O)C(CC)c1ccccc1
<b>Mol. weight [g/mol]:</b>	274.40

## Physical Properties

Property code	Value	Unit	Source
gf	54.51	kJ/mol	Joback Method
hf	-316.46	kJ/mol	Joback Method
hfus	32.36	kJ/mol	Joback Method
hvap	66.28	kJ/mol	Joback Method
log10ws	-5.25		Crippen Method
logp	4.858		Crippen Method
mvol	243.860	ml/mol	McGowan Method
pc	1598.72	kPa	Joback Method
rinpol	1806.00		NIST Webbook
rinpol	1806.00		NIST Webbook
tb	717.49	K	Joback Method
tc	921.71	K	Joback Method
tf	356.12	K	Joback Method
vc	0.927	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	688.78	J/molxK	717.49	Joback Method
cpg	767.86	J/molxK	887.68	Joback Method
cpg	754.03	J/molxK	853.64	Joback Method
cpg	739.26	J/molxK	819.60	Joback Method
cpg	723.49	J/molxK	785.56	Joback Method
cpg	706.68	J/molxK	751.53	Joback Method
cpg	780.81	J/molxK	921.71	Joback Method
dvisc	0.0000707	Paxs	717.49	Joback Method

dvisc	0.0000968	Paxs	657.26	Joback Method
dvisc	0.0001411	Paxs	597.03	Joback Method
dvisc	0.0002240	Paxs	536.81	Joback Method
dvisc	0.0003997	Paxs	476.58	Joback Method
dvisc	0.0008430	Paxs	416.35	Joback Method
dvisc	0.0022889	Paxs	356.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406917&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406917&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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