

# Succinic acid, 3-methylbut-2-yl 2-isopropoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C18H26O5/c1-12(2)14(5)22-17(19)10-11-18(20)23-16-9-7-6-8-15(16)21-13(3)
<b>InchiKey:</b>	NGLCUALXIUIYSH-UHFFFAOYSA-N
<b>Formula:</b>	C18H26O5
<b>SMILES:</b>	CC(C)Oc1ccccc1OC(=O)CCC(=O)OC(C)C(C)C
<b>Mol. weight [g/mol]:</b>	322.40

## Physical Properties

Property code	Value	Unit	Source
gf	-376.70	kJ/mol	Joback Method
hf	-827.45	kJ/mol	Joback Method
hfus	32.22	kJ/mol	Joback Method
hvap	78.16	kJ/mol	Joback Method
log10ws	-4.52		Crippen Method
logp	3.747		Crippen Method
mvol	261.470	ml/mol	McGowan Method
pc	1561.05	kPa	Joback Method
rinpol	2121.00		NIST Webbook
rinpol	2121.00		NIST Webbook
tb	816.58	K	Joback Method
tc	1023.31	K	Joback Method
tf	453.11	K	Joback Method
vc	0.984	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	791.62	J/molxK	816.58	Joback Method
cpg	807.35	J/molxK	851.03	Joback Method
cpg	821.88	J/molxK	885.49	Joback Method
cpg	835.21	J/molxK	919.94	Joback Method
cpg	847.33	J/molxK	954.40	Joback Method
cpg	858.27	J/molxK	988.85	Joback Method
cpg	868.01	J/molxK	1023.31	Joback Method
dvisc	0.0007705	Paxs	453.11	Joback Method

dvisc	0.0003567	Paxs	513.69	Joback Method
dvisc	0.0001942	Paxs	574.27	Joback Method
dvisc	0.0001188	Paxs	634.85	Joback Method
dvisc	0.0000791	Paxs	695.42	Joback Method
dvisc	0.0000563	Paxs	756.00	Joback Method
dvisc	0.0000421	Paxs	816.58	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389787&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389787&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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