

# Succinic acid, hexyl 2-phenoxyethyl ester

<b>Inchi:</b>	InChI=1S/C18H26O5/c1-2-3-4-8-13-22-17(19)11-12-18(20)23-15-14-21-16-9-6-5-7-10-16
<b>InchiKey:</b>	BCWSHSUAFAMZNF-UHFFFAOYSA-N
<b>Formula:</b>	C18H26O5
<b>SMILES:</b>	CCCCCOC(=O)CCC(=O)OCCOc1ccccc1
<b>Mol. weight [g/mol]:</b>	322.40

## Physical Properties

Property code	Value	Unit	Source
gf	-359.75	kJ/mol	Joback Method
hf	-800.14	kJ/mol	Joback Method
hfus	43.18	kJ/mol	Joback Method
hvap	78.66	kJ/mol	Joback Method
log10ws	-3.92		Crippen Method
logp	3.512		Crippen Method
mcvol	261.470	ml/mol	McGowan Method
pc	1550.00	kPa	Joback Method
rinpol	2359.00		NIST Webbook
rinpol	2359.00		NIST Webbook
tb	812.92	K	Joback Method
tc	1012.09	K	Joback Method
tf	485.59	K	Joback Method
vc	1.002	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	790.76	J/molxK	812.92	Joback Method
cpg	806.10	J/molxK	846.11	Joback Method
cpg	820.33	J/molxK	879.31	Joback Method
cpg	833.48	J/molxK	912.50	Joback Method
cpg	845.54	J/molxK	945.70	Joback Method
cpg	856.54	J/molxK	978.89	Joback Method
cpg	866.47	J/molxK	1012.09	Joback Method
dvisc	0.0005959	Paxs	485.59	Joback Method

dvisc	0.0003262	Paxs	540.14	Joback Method
dvisc	0.0001994	Paxs	594.70	Joback Method
dvisc	0.0001324	Paxs	649.25	Joback Method
dvisc	0.0000937	Paxs	703.81	Joback Method
dvisc	0.0000697	Paxs	758.37	Joback Method
dvisc	0.0000539	Paxs	812.92	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381197&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381197&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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