

# Succinic acid, cyclohexylmethyl 4-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C18H24O5/c1-21-15-7-9-16(10-8-15)23-18(20)12-11-17(19)22-13-14-5-3-2-4-6
<b>InchiKey:</b>	PZNAQQQSPKCTKQ-UHFFFAOYSA-N
<b>Formula:</b>	C18H24O5
<b>SMILES:</b>	COc1ccc(OC(=O)CCC(=O)OCC2CCCCC2)cc1
<b>Mol. weight [g/mol]:</b>	320.38

## Physical Properties

Property code	Value	Unit	Source
gf	-344.93	kJ/mol	Joback Method
hf	-757.29	kJ/mol	Joback Method
hfus	34.62	kJ/mol	Joback Method
hvap	79.75	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	3.504		Crippen Method
mvol	250.610	ml/mol	McGowan Method
pc	1810.77	kPa	Joback Method
rinpol	2583.00		NIST Webbook
rinpol	2583.00		NIST Webbook
tb	837.45	K	Joback Method
tc	1059.18	K	Joback Method
tf	505.49	K	Joback Method
vc	0.934	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	784.54	J/molxK	837.45	Joback Method
cpg	800.83	J/molxK	874.40	Joback Method
cpg	815.61	J/molxK	911.36	Joback Method
cpg	828.91	J/molxK	948.31	Joback Method
cpg	840.72	J/molxK	985.27	Joback Method
cpg	851.05	J/molxK	1022.22	Joback Method
cpg	859.93	J/molxK	1059.18	Joback Method
dvisc	0.0005796	Paxs	505.49	Joback Method

dvisc	0.0003246	Paxs	560.82	Joback Method
dvisc	0.0002017	Paxs	616.14	Joback Method
dvisc	0.0001356	Paxs	671.47	Joback Method
dvisc	0.0000968	Paxs	726.80	Joback Method
dvisc	0.0000725	Paxs	782.12	Joback Method
dvisc	0.0000564	Paxs	837.45	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389829&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389829&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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