

# Fumaric acid, 3-phenylpropyl 3-methylbut-2-yl ester

Inchi:	InChI=1S/C18H24O4/c1-14(2)15(3)22-18(20)12-11-17(19)21-13-7-10-16-8-5-4-6-9-16/h
InchiKey:	ZHMLFTLVMAJOHS-VAWYXSNFSA-N
Formula:	C18H24O4
SMILES:	CC(C)C(C)OC(=O)C=CC(=O)OCCc1ccccc1
Mol. weight [g/mol]:	304.38

## Physical Properties

Property code	Value	Unit	Source
gf	-179.41	kJ/mol	Joback Method
hf	-561.26	kJ/mol	Joback Method
hfus	35.15	kJ/mol	Joback Method
hvap	75.43	kJ/mol	Joback Method
log10ws	-3.91		Crippen Method
logp	3.306		Crippen Method
mvol	251.300	ml/mol	McGowan Method
pc	1659.19	kPa	Joback Method
rinpol	2190.00		NIST Webbook
rinpol	2190.00		NIST Webbook
tb	793.78	K	Joback Method
tc	1003.25	K	Joback Method
tf	428.28	K	Joback Method
vc	0.952	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	737.65	J/molxK	793.78	Joback Method
cpg	753.32	J/molxK	828.69	Joback Method
cpg	767.89	J/molxK	863.60	Joback Method
cpg	781.40	J/molxK	898.52	Joback Method
cpg	793.88	J/molxK	933.43	Joback Method
cpg	805.37	J/molxK	968.34	Joback Method
cpg	815.92	J/molxK	1003.25	Joback Method
dvisc	0.0010956	Paxs	428.28	Joback Method

dvisc	0.0004816	Paxs	489.20	Joback Method
dvisc	0.0002539	Paxs	550.11	Joback Method
dvisc	0.0001521	Paxs	611.03	Joback Method
dvisc	0.0001000	Paxs	671.95	Joback Method
dvisc	0.0000705	Paxs	732.86	Joback Method
dvisc	0.0000524	Paxs	793.78	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405661&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405661&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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