

Hydratropic acid, 2-methyloct-5-yn-4-yl ester

Inchi:

lnchl=1S/C18H24O2/c1-5-6-12-17(13-14(2)3)20-18(19)15(4)16-10-8-7-9-11-16/h7-11,14

InchiKey:

HXVNSFXUARBLNQ-UHFFFAOYSA-N

Formula:

C18H24O2

SMILES:

CCC#CC(CC(C)C)OC(=O)C(C)c1ccccc1

Mol. weight [g/mol]:

272.38

Physical Properties

Property code	Value	Unit	Source
gf	174.65	kJ/mol	Joback Method
hf	-166.66	kJ/mol	Joback Method
hfus	31.76	kJ/mol	Joback Method
hvap	68.08	kJ/mol	Joback Method
log10ws	-4.95		Crippen Method
logp	4.161		Crippen Method
mcvol	239.560	ml/mol	McGowan Method
pc	1758.02	kPa	Joback Method
rinpol	1793.00		NIST Webbook
rinpol	1793.00		NIST Webbook
tb	721.89	K	Joback Method
tc	942.19	K	Joback Method
tf	452.30	K	Joback Method
vc	0.903	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	667.00	J/mol×K	721.89	Joback Method
cpg	685.32	J/mol×K	758.61	Joback Method
cpg	702.43	J/mol×K	795.32	Joback Method
cpg	718.36	J/mol×K	832.04	Joback Method
cpg	733.16	J/mol×K	868.76	Joback Method
cpg	746.87	J/mol×K	905.48	Joback Method
cpg	759.53	J/mol×K	942.19	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U406958&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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