

# Cyclopropanecarboxylic acid, trans-2-phenyl-, oct-3-en-2-yl ester

Inchi:	InChI=1S/C18H24O2/c1-3-4-5-7-10-14(2)20-18(19)17-13-16(17)15-11-8-6-9-12-15/h6-12
InchiKey:	YMMIEKBQBRAIEL-JXMROGBWSA-N
Formula:	C18H24O2
SMILES:	CCCCC=CC(C)OC(=O)C1CC1c1ccccc1
Mol. weight [g/mol]:	272.38

## Physical Properties

Property code	Value	Unit	Source
gf	109.99	kJ/mol	Joback Method
hf	-258.72	kJ/mol	Joback Method
hfus	35.09	kJ/mol	Joback Method
hvap	66.27	kJ/mol	Joback Method
log10ws	-4.90		Crippen Method
logp	4.468		Crippen Method
mcvol	233.000	ml/mol	McGowan Method
pc	1718.88	kPa	Joback Method
rinpol	2032.00		NIST Webbook
rinpol	2032.00		NIST Webbook
tb	720.00	K	Joback Method
tc	932.25	K	Joback Method
tf	384.82	K	Joback Method
vc	0.889	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	675.77	J/molxK	720.00	Joback Method
cpg	694.16	J/molxK	755.37	Joback Method
cpg	711.38	J/molxK	790.75	Joback Method
cpg	727.50	J/molxK	826.12	Joback Method
cpg	742.60	J/molxK	861.50	Joback Method
cpg	756.75	J/molxK	896.87	Joback Method
cpg	770.03	J/molxK	932.25	Joback Method
dvisc	0.0019315	Paxs	384.82	Joback Method

dvisc	0.0011188	Paxs	440.68	Joback Method
dvisc	0.0007328	Paxs	496.55	Joback Method
dvisc	0.0005229	Paxs	552.41	Joback Method
dvisc	0.0003969	Paxs	608.27	Joback Method
dvisc	0.0003156	Paxs	664.14	Joback Method
dvisc	0.0002601	Paxs	720.00	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406909&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406909&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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