

# 2-(2-(2-(2-(2-Isopentoxy-ethoxy)-ethoxy)-ethoxy)-ethoxy)-ethoxy)-acetate

Inchi:  
InchiKey:

InChI=1S/C19H38O8/c1-18(2)4-5-21-6-7-22-8-9-23-10-11-24-12-13-25-14-15-26-16-17-2

CPWGIFFQMSJIMU-UHFFFAOYSA-N

Formula:

C19H38O8

SMILES:

CC(=O)OCCOCCOCCOCCOCCOCCOCCOCC(C)C

Mol. weight [g/mol]:

394.50

## Physical Properties

Property code	Value	Unit	Source
gf	-757.26	kJ/mol	Joback Method
hf	-1478.89	kJ/mol	Joback Method
hfus	51.36	kJ/mol	Joback Method
hvap	81.12	kJ/mol	Joback Method
log10ws	-0.92		Crippen Method
logp	1.695		Crippen Method
mcvol	321.230	ml/mol	McGowan Method
pc	1047.33	kPa	Joback Method
rinpol	2592.80		NIST Webbook
rinpol	2592.80		NIST Webbook
tb	844.49	K	Joback Method
tc	1033.91	K	Joback Method
tf	494.43	K	Joback Method
vc	1.226	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1042.78	J/molxK	844.49	Joback Method
cpg	1060.92	J/molxK	876.06	Joback Method
cpg	1077.70	J/molxK	907.63	Joback Method
cpg	1093.07	J/molxK	939.20	Joback Method
cpg	1107.01	J/molxK	970.77	Joback Method
cpg	1119.49	J/molxK	1002.34	Joback Method
cpg	1130.46	J/molxK	1033.91	Joback Method
dvisc	0.0002069	Paxs	494.43	Joback Method

dvisc	0.0001037	Paxs	552.77	Joback Method
dvisc	0.0000593	Paxs	611.12	Joback Method
dvisc	0.0000373	Paxs	669.46	Joback Method
dvisc	0.0000253	Paxs	727.80	Joback Method
dvisc	0.0000182	Paxs	786.15	Joback Method
dvisc	0.0000137	Paxs	844.49	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R188151&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R188151&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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