

# Fumaric acid, decyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C19H32O4/c1-3-5-7-8-9-10-11-13-17-23-19(21)15-14-18(20)22-16-12-6-4-2/h
<b>InchiKey:</b>	NRJKRQBIPXJFTO-CCEZHUSRSA-N
<b>Formula:</b>	C19H32O4
<b>SMILES:</b>	<chem>C=CCCCOC(=O)C=CC(=O)OCCCCCCCCC</chem>
<b>Mol. weight [g/mol]:</b>	324.45

## Physical Properties

Property code	Value	Unit	Source
gf	-190.68	kJ/mol	Joback Method
hf	-682.44	kJ/mol	Joback Method
hfus	49.46	kJ/mol	Joback Method
hvap	75.49	kJ/mol	Joback Method
log10ws	-5.21		Crippen Method
logp	4.736		Crippen Method
mvol	284.850	ml/mol	McGowan Method
pc	1225.98	kPa	Joback Method
rinpol	2284.00		NIST Webbook
rinpol	2284.00		NIST Webbook
tb	787.54	K	Joback Method
tc	972.17	K	Joback Method
tf	441.37	K	Joback Method
vc	1.109	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	857.70	J/molxK	787.54	Joback Method
cpg	932.51	J/molxK	941.40	Joback Method
cpg	919.28	J/molxK	910.63	Joback Method
cpg	905.20	J/molxK	879.86	Joback Method
cpg	890.27	J/molxK	849.08	Joback Method
cpg	874.44	J/molxK	818.31	Joback Method
cpg	944.94	J/molxK	972.17	Joback Method
dvisc	0.0000568	Paxs	787.54	Joback Method

dvisc	0.0000749	Paxs	729.85	Joback Method
dvisc	0.0001034	Paxs	672.15	Joback Method
dvisc	0.0001518	Paxs	614.46	Joback Method
dvisc	0.0002412	Paxs	556.76	Joback Method
dvisc	0.0004268	Paxs	499.07	Joback Method
dvisc	0.0008763	Paxs	441.37	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348851&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348851&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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