

# 1,2-Cyclohexanedicarboxylic acid, ethyl hexyl ester

Inchi:	InChI=1S/C16H28O4/c1-3-5-6-9-12-20-16(18)14-11-8-7-10-13(14)15(17)19-4-2/h13-14H
InchiKey:	DNFPNUCSOCQYSZ-UHFFFAOYSA-N
Formula:	C16H28O4
SMILES:	CCCCCOC(=O)C1CCCCC1C(=O)OCC
Mol. weight [g/mol]:	284.39

## Physical Properties

Property code	Value	Unit	Source
gf	-367.26	kJ/mol	Joback Method
hf	-829.19	kJ/mol	Joback Method
hfus	35.68	kJ/mol	Joback Method
hvap	69.64	kJ/mol	Joback Method
log10ws	-3.66		Crippen Method
logp	3.479		Crippen Method
mvol	240.320	ml/mol	McGowan Method
pc	1612.88	kPa	Joback Method
rinpol	1935.00		NIST Webbook
rinpol	1935.00		NIST Webbook
tb	732.94	K	Joback Method
tc	929.01	K	Joback Method
tf	417.54	K	Joback Method
vc	0.911	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	731.43	J/molxK	732.94	Joback Method
cpg	813.84	J/molxK	896.33	Joback Method
cpg	799.62	J/molxK	863.65	Joback Method
cpg	784.28	J/molxK	830.97	Joback Method
cpg	767.80	J/molxK	798.30	Joback Method
cpg	750.19	J/molxK	765.62	Joback Method
cpg	826.93	J/molxK	929.01	Joback Method
dvisc	0.0001198	Paxs	732.94	Joback Method

dvisc	0.0001547	Paxs	680.37	Joback Method
dvisc	0.0002087	Paxs	627.81	Joback Method
dvisc	0.0002972	Paxs	575.24	Joback Method
dvisc	0.0004546	Paxs	522.67	Joback Method
dvisc	0.0007645	Paxs	470.11	Joback Method
dvisc	0.0014656	Paxs	417.54	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339406&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339406&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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