

# Acetanilide, 2,6-diethyl-4-tert-butyl-

Inchi:	lnChI=1S/C16H25NO/c1-7-12-9-14(16(4,5)6)10-13(8-2)15(12)17-11(3)18/h9-10H,7-8H2
InchiKey:	OSSFCCXGYSEKLZ-UHFFFAOYSA-N
Formula:	C16H25NO
SMILES:	CCc1cc(C(C)(C)C)cc(CC)c1N=C(C)O
Mol. weight [g/mol]:	247.38

## Physical Properties

Property code	Value	Unit	Source
hf	-260.00	kJ/mol	Joback Method
hvap	74.25	kJ/mol	Joback Method
log10ws	-4.78		Crippen Method
logp	4.717		Crippen Method
mcvol	224.090	ml/mol	McGowan Method
pc	1656.49	kPa	Joback Method
tb	772.61	K	Joback Method
tc	979.95	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=B6009293&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=B6009293&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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