

# Cyclohexanecarboxylic acid, 4-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C14H18O3/c1-16-12-7-9-13(10-8-12)17-14(15)11-5-3-2-4-6-11/h7-11H,2-6H2,
<b>InchiKey:</b>	FRLVVPHULLQDHW-UHFFFAOYSA-N
<b>Formula:</b>	C14H18O3
<b>SMILES:</b>	COc1ccc(OC(=O)C2CCCCC2)cc1
<b>Mol. weight [g/mol]:</b>	234.29

## Physical Properties

Property code	Value	Unit	Source
gf	-144.69	kJ/mol	Joback Method
hf	-429.93	kJ/mol	Joback Method
hfus	21.48	kJ/mol	Joback Method
hvap	61.69	kJ/mol	Joback Method
log10ws	-3.65		Crippen Method
logp	3.181		Crippen Method
mvol	186.810	ml/mol	McGowan Method
pc	2470.27	kPa	Joback Method
rinpol	1862.00		NIST Webbook
rinpol	1862.00		NIST Webbook
tb	669.64	K	Joback Method
tc	902.28	K	Joback Method
tf	388.25	K	Joback Method
vc	0.686	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	513.12	J/molxK	669.64	Joback Method
cpg	593.57	J/molxK	863.50	Joback Method
cpg	580.08	J/molxK	824.73	Joback Method
cpg	565.30	J/molxK	785.96	Joback Method
cpg	549.23	J/molxK	747.19	Joback Method
cpg	531.84	J/molxK	708.41	Joback Method
cpg	605.80	J/molxK	902.28	Joback Method
dvisc	0.0001251	Paxs	669.64	Joback Method

dvisc	0.0001607	Paxs	622.74	Joback Method
dvisc	0.0002150	Paxs	575.84	Joback Method
dvisc	0.0003029	Paxs	528.95	Joback Method
dvisc	0.0004562	Paxs	482.05	Joback Method
dvisc	0.0007502	Paxs	435.15	Joback Method
dvisc	0.0013915	Paxs	388.25	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307705&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307705&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/89-842-9/Cyclohexanecarboxylic-acid-4-methoxyphenyl-ester.pdf>

Generated by Cheméo on 2024-04-24 07:47:46.760366225 +0000 UTC m=+16234115.680943537.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.