

# 2-[1-(4-Cyano-1,2,3,4-tetrahydronaphthyl)]propane

<b>Other names:</b>	4-(1-Cyanoethyl)-1,2,3,4-tetrahydro-1-naphthalenecarbonitrile
<b>Inchi:</b>	InChI=1S/C14H14N2/c1-10(8-15)12-7-6-11(9-16)13-4-2-3-5-14(12)13/h2-5,10-12H,6-7H
<b>InchiKey:</b>	FDTQTOKRWQJRAC-UHFFFAOYSA-N
<b>Formula:</b>	C14H14N2
<b>SMILES:</b>	CC(C#N)C1CCC(C#N)c2ccccc21
<b>Mol. weight [g/mol]:</b>	210.27
<b>CAS:</b>	57964-39-3

## Physical Properties

Property code	Value	Unit	Source
gf	474.64	kJ/mol	Joback Method
hf	263.55	kJ/mol	Joback Method
hfus	22.26	kJ/mol	Joback Method
hvap	70.04	kJ/mol	Joback Method
log10ws	-4.07		Crippen Method
logp	3.331		Crippen Method
mcvol	176.260	ml/mol	McGowan Method
pc	2131.49	kPa	Joback Method
rinpol	2078.00		NIST Webbook
rinpol	2078.00		NIST Webbook
rinpol	2078.00		NIST Webbook
tb	761.44	K	Joback Method
tc	1007.47	K	Joback Method
tf	411.64	K	Joback Method
vc	0.706	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	496.03	J/molxK	761.44	Joback Method
cpg	509.67	J/molxK	802.44	Joback Method
cpg	522.23	J/molxK	843.45	Joback Method
cpg	533.80	J/molxK	884.45	Joback Method
cpg	544.45	J/molxK	925.46	Joback Method

cpg	554.26	J/mol×K	966.46	Joback Method
cpg	563.30	J/mol×K	1007.47	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C57964393&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C57964393&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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