

Diethyl isobutylmalonate

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| Other names: | Ethyl 2-carbethoxy-4-methyl pentanoate Ethyl isobutyl malonate Isobutylmalonic acid diethyl ester Propanedioic acid, (2-methylpropyl)-, diethyl ester Malonic acid, isobutyl-, diethyl ester |
| Inchi: | InChI=1S/C11H20O4/c1-5-14-10(12)9(7-8(3)4)11(13)15-6-2/h8-9H,5-7H2,1-4H3 |
| InchiKey: | OFRFGNSZCYDFOH-UHFFFAOYSA-N |
| Formula: | C11H20O4 |
| SMILES: | CCOC(=O)C(CC(C)C)C(=O)OCC |
| Mol. weight [g/mol]: | 216.27 |
| CAS: | 10203-58-4 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -430.98 | kJ/mol | Joback Method |
| hf | -770.53 | kJ/mol | Joback Method |
| hfus | 22.77 | kJ/mol | Joback Method |
| hvap | 57.62 | kJ/mol | Joback Method |
| log10ws | -1.67 | | Crippen Method |
| logp | 1.775 | | Crippen Method |
| mcvol | 180.730 | ml/mol | McGowan Method |
| pc | 2131.49 | kPa | Joback Method |
| tb | 602.78 | K | Joback Method |
| tc | 787.66 | K | Joback Method |
| tf | 328.05 | K | Joback Method |
| vc | 0.688 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 465.83 | J/molxK | 602.78 | Joback Method |
| cpg | 531.86 | J/molxK | 756.85 | Joback Method |
| cpg | 519.94 | J/molxK | 726.03 | Joback Method |
| cpg | 507.38 | J/molxK | 695.22 | Joback Method |

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|-------|-----------|---------|--------|---------------|
| cpg | 494.18 | J/molxK | 664.41 | Joback Method |
| cpg | 480.33 | J/molxK | 633.59 | Joback Method |
| cpg | 543.12 | J/molxK | 787.66 | Joback Method |
| dvisc | 0.0001484 | Paxs | 602.78 | Joback Method |
| dvisc | 0.0001992 | Paxs | 556.99 | Joback Method |
| dvisc | 0.0002818 | Paxs | 511.20 | Joback Method |
| dvisc | 0.0004269 | Paxs | 465.41 | Joback Method |
| dvisc | 0.0007082 | Paxs | 419.63 | Joback Method |
| dvisc | 0.0013297 | Paxs | 373.84 | Joback Method |
| dvisc | 0.0029767 | Paxs | 328.05 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C10203584&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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