

# 1,3-Dioxane, 2-isopentyl-4-(2-pentenyl), 2S,4R

<b>Inchi:</b>	InChI=1S/C14H26O2/c1-4-5-6-7-13-10-11-15-14(16-13)9-8-12(2)3/h5-6,12-14H,4,7-11H2
<b>InchiKey:</b>	LZKQPIVTALJJKY-BUHQSOCUSA-N
<b>Formula:</b>	C14H26O2
<b>SMILES:</b>	CCC=CCC1CCOC(CCC(C)C)O1
<b>Mol. weight [g/mol]:</b>	226.35

## Physical Properties

Property code	Value	Unit	Source
gf	-10.72	kJ/mol	Joback Method
hf	-450.37	kJ/mol	Joback Method
hfus	37.56	kJ/mol	Joback Method
hvap	55.47	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	3.910		Crippen Method
mvol	204.700	ml/mol	McGowan Method
pc	1813.86	kPa	Joback Method
ripol	1663.00		NIST Webbook
ripol	1663.00		NIST Webbook
tb	592.22	K	Joback Method
tc	789.59	K	Joback Method
tf	283.74	K	Joback Method
vc	0.767	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	553.65	J/molxK	592.22	Joback Method
cpg	574.47	J/molxK	625.11	Joback Method
cpg	594.18	J/molxK	658.01	Joback Method
cpg	612.82	J/molxK	690.90	Joback Method
cpg	630.43	J/molxK	723.80	Joback Method
cpg	647.04	J/molxK	756.69	Joback Method
cpg	662.69	J/molxK	789.59	Joback Method
dvisc	0.0062441	Paxs	283.74	Joback Method

dvisc	0.0021488	Paxs	335.15	Joback Method
dvisc	0.0009821	Paxs	386.57	Joback Method
dvisc	0.0005394	Paxs	437.98	Joback Method
dvisc	0.0003361	Paxs	489.39	Joback Method
dvisc	0.0002291	Paxs	540.81	Joback Method
dvisc	0.0001669	Paxs	592.22	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R191870&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R191870&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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