

# Benzene, 1,3-dimethoxy-2-hexyl

<b>Inchi:</b>	InChI=1S/C14H22O2/c1-4-5-6-7-9-12-13(15-2)10-8-11-14(12)16-3/h8,10-11H,4-7,9H2,1-
<b>InchiKey:</b>	ORCSRUIWGMBCRCQ-UHFFFAOYSA-N
<b>Formula:</b>	C14H22O2
<b>SMILES:</b>	CCCCCCc1c(OC)cccc1OC
<b>Mol. weight [g/mol]:</b>	222.32

## Physical Properties

Property code	Value	Unit	Source
gf	-49.85	kJ/mol	Joback Method
hf	-383.14	kJ/mol	Joback Method
hfus	27.65	kJ/mol	Joback Method
hvap	55.18	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	3.827		Crippen Method
mcvol	196.100	ml/mol	McGowan Method
pc	1898.60	kPa	Joback Method
rinpol	1610.00		NIST Webbook
rinpol	1610.00		NIST Webbook
tb	601.20	K	Joback Method
tc	794.21	K	Joback Method
tf	343.46	K	Joback Method
vc	0.748	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	498.23	J/molxK	601.20	Joback Method
cpg	574.91	J/molxK	762.04	Joback Method
cpg	561.10	J/molxK	729.87	Joback Method
cpg	546.53	J/molxK	697.71	Joback Method
cpg	531.19	J/molxK	665.54	Joback Method
cpg	515.10	J/molxK	633.37	Joback Method
cpg	587.97	J/molxK	794.21	Joback Method
dvisc	0.0001126	Paxs	601.20	Joback Method

dvisc	0.0001419	Paxs	558.24	Joback Method
dvisc	0.0001857	Paxs	515.29	Joback Method
dvisc	0.0002553	Paxs	472.33	Joback Method
dvisc	0.0003741	Paxs	429.37	Joback Method
dvisc	0.0005968	Paxs	386.42	Joback Method
dvisc	0.0010700	Paxs	343.46	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R142853&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R142853&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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