

# Isophthalic acid, ethyl 3-methylphenyl ester

<b>Inchi:</b>	InChI=1S/C17H16O4/c1-3-20-16(18)13-7-5-8-14(11-13)17(19)21-15-9-4-6-12(2)10-15/h4
<b>InchiKey:</b>	OFKGOZKVKPWQAY-UHFFFAOYSA-N
<b>Formula:</b>	C17H16O4
<b>SMILES:</b>	CCOC(=O)c1cccc(C(=O)Oc2cccc(C)c2)c1
<b>Mol. weight [g/mol]:</b>	284.31

## Physical Properties

Property code	Value	Unit	Source
gf	-170.02	kJ/mol	Joback Method
hf	-433.69	kJ/mol	Joback Method
hfus	32.66	kJ/mol	Joback Method
hvap	77.62	kJ/mol	Joback Method
log10ws	-4.70		Crippen Method
logp	3.391		Crippen Method
mvol	217.750	ml/mol	McGowan Method
pc	2212.45	kPa	Joback Method
rinpol	2362.00		NIST Webbook
rinpol	2362.00		NIST Webbook
tb	804.26	K	Joback Method
tc	1036.96	K	Joback Method
tf	503.55	K	Joback Method
vc	0.820	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	611.57	J/molxK	804.26	Joback Method
cpg	667.89	J/molxK	998.17	Joback Method
cpg	658.98	J/molxK	959.39	Joback Method
cpg	648.92	J/molxK	920.61	Joback Method
cpg	637.67	J/molxK	881.83	Joback Method
cpg	625.23	J/molxK	843.04	Joback Method
cpg	675.67	J/molxK	1036.96	Joback Method
dvisc	0.0000843	Paxs	804.26	Joback Method

dvisc	0.0001046	Paxs	754.14	Joback Method
dvisc	0.0001339	Paxs	704.02	Joback Method
dvisc	0.0001779	Paxs	653.90	Joback Method
dvisc	0.0002478	Paxs	603.79	Joback Method
dvisc	0.0003666	Paxs	553.67	Joback Method
dvisc	0.0005863	Paxs	503.55	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344514&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344514&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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