

# 1,2-Cyclohexanedicarboxylic acid, isohexyl isopropyl ester

<b>Inchi:</b>	InChI=1S/C17H30O4/c1-12(2)8-7-11-20-16(18)14-9-5-6-10-15(14)17(19)21-13(3)4/h12-1
<b>InchiKey:</b>	ONUIWTQGQPQZMLW-UHFFFAOYSA-N
<b>Formula:</b>	C17H30O4
<b>SMILES:</b>	CC(C)CCCOC(=O)C1CCCCC1C(=O)OC(C)C
<b>Mol. weight [g/mol]:</b>	298.42

## Physical Properties

Property code	Value	Unit	Source
gf	-363.72	kJ/mol	Joback Method
hf	-860.39	kJ/mol	Joback Method
hfus	31.22	kJ/mol	Joback Method
hvap	71.09	kJ/mol	Joback Method
log10ws	-3.95		Crippen Method
logp	3.724		Crippen Method
mvol	254.410	ml/mol	McGowan Method
pc	1514.03	kPa	Joback Method
rinpol	2102.00		NIST Webbook
rinpol	2102.00		NIST Webbook
tb	754.94	K	Joback Method
tc	955.18	K	Joback Method
tf	398.81	K	Joback Method
vc	0.956	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	790.61	J/mol×K	754.94	Joback Method
cpg	810.05	J/mol×K	788.31	Joback Method
cpg	828.22	J/mol×K	821.69	Joback Method
cpg	845.14	J/mol×K	855.06	Joback Method
cpg	860.80	J/mol×K	888.43	Joback Method
cpg	875.23	J/mol×K	921.80	Joback Method
cpg	888.42	J/mol×K	955.18	Joback Method
dvisc	0.0019397	Paxs	398.81	Joback Method

dvisc	0.0008336	Paxs	458.17	Joback Method
dvisc	0.0004348	Paxs	517.52	Joback Method
dvisc	0.0002593	Paxs	576.88	Joback Method
dvisc	0.0001703	Paxs	636.23	Joback Method
dvisc	0.0001202	Paxs	695.59	Joback Method
dvisc	0.0000896	Paxs	754.94	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339639&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339639&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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