

# Fumaric acid, 2,4,4-trimethylpentyl 3-methylbut-2-yl ester

<b>Inchi:</b>	InChI=1S/C17H30O4/c1-12(2)14(4)21-16(19)9-8-15(18)20-11-13(3)10-17(5,6)7/h8-9,12-
<b>InchiKey:</b>	ULINRDVYPHZRRN-CMDGGGOBGSA-N
<b>Formula:</b>	C17H30O4
<b>SMILES:</b>	CC(COC(=O)C=CC(=O)OC(C)C(C)C)CC(C)(C)C
<b>Mol. weight [g/mol]:</b>	298.42

## Physical Properties

Property code	Value	Unit	Source
gf	-299.84	kJ/mol	Joback Method
hf	-791.18	kJ/mol	Joback Method
hfus	27.58	kJ/mol	Joback Method
hvap	69.25	kJ/mol	Joback Method
log10ws	-3.90		Crippen Method
logp	3.746		Crippen Method
mcvol	260.970	ml/mol	McGowan Method
pc	1410.13	kPa	Joback Method
rinpol	1830.00		NIST Webbook
rinpol	1830.00		NIST Webbook
tb	740.55	K	Joback Method
tc	934.01	K	Joback Method
tf	378.01	K	Joback Method
vc	0.987	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	770.66	J/molxK	740.55	Joback Method
cpg	847.81	J/molxK	901.76	Joback Method
cpg	834.26	J/molxK	869.52	Joback Method
cpg	819.80	J/molxK	837.28	Joback Method
cpg	804.41	J/molxK	805.04	Joback Method
cpg	788.04	J/molxK	772.79	Joback Method
cpg	860.49	J/molxK	934.01	Joback Method
dvisc	0.0000453	Paxs	740.55	Joback Method

dvisc	0.0000648	Paxs	680.13	Joback Method
dvisc	0.0000993	Paxs	619.70	Joback Method
dvisc	0.0001669	Paxs	559.28	Joback Method
dvisc	0.0003182	Paxs	498.86	Joback Method
dvisc	0.0007244	Paxs	438.43	Joback Method
dvisc	0.0021453	Paxs	378.01	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405601&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405601&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/90-235-1/Fumaric-acid-2-4-4-trimethylpentyl-3-methylbut-2-yl-ester.pdf>

Generated by Cheméo on 2024-04-25 18:16:59.562552659 +0000 UTC m=+16358268.483129971.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.