

# 1,2-Cyclohexanedicarboxylic acid, isoheptyl 2-methoxyethyl ester

**Inchi:** InChI=1S/C17H30O5/c1-13(2)7-6-10-21-16(18)14-8-4-5-9-15(14)17(19)22-12-11-20-3/h1-13,15-19,21-22  
**InchiKey:** ZFUVEGUFQGMEN-UHFFFAOYSA-N  
**Formula:** C17H30O5  
**SMILES:** COCCOC(=O)C1CCCCC1C(=O)OCCCC(C)C  
**Mol. weight [g/mol]:** 314.42

## Physical Properties

Property code	Value	Unit	Source
gf	-466.28	kJ/mol	Joback Method
hf	-987.33	kJ/mol	Joback Method
hfus	35.93	kJ/mol	Joback Method
hvap	73.89	kJ/mol	Joback Method
log10ws	-2.92		Crippen Method
logp	2.962		Crippen Method
mvol	260.280	ml/mol	McGowan Method
pc	1485.00	kPa	Joback Method
rinpol	2089.00		NIST Webbook
rinpol	2089.00		NIST Webbook
tb	777.80	K	Joback Method
tc	975.15	K	Joback Method
tf	436.04	K	Joback Method
vc	0.980	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	821.02	J/molxK	777.80	Joback Method
cpg	901.48	J/molxK	942.26	Joback Method
cpg	887.93	J/molxK	909.36	Joback Method
cpg	873.10	J/molxK	876.47	Joback Method
cpg	857.01	J/molxK	843.58	Joback Method
cpg	839.65	J/molxK	810.69	Joback Method
cpg	913.75	J/molxK	975.15	Joback Method
dvisc	0.0000739	Paxs	777.80	Joback Method

dvisc	0.0000970	Paxs	720.84	Joback Method
dvisc	0.0001335	Paxs	663.88	Joback Method
dvisc	0.0001952	Paxs	606.92	Joback Method
dvisc	0.0003086	Paxs	549.96	Joback Method
dvisc	0.0005423	Paxs	493.00	Joback Method
dvisc	0.0011045	Paxs	436.04	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U340027&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U340027&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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