

# Cyclopropanecarboxylic acid, trans-2-phenyl-, hept-2-yl ester

<b>Inchi:</b>	InChI=1S/C17H24O2/c1-3-4-6-9-13(2)19-17(18)16-12-15(16)14-10-7-5-8-11-14/h5,7-8,1
<b>InchiKey:</b>	SINSXGWSOOGAOP-UHFFFAOYSA-N
<b>Formula:</b>	C17H24O2
<b>SMILES:</b>	CCCCC(C)OC(=O)C1CC1c1ccccc1
<b>Mol. weight [g/mol]:</b>	260.37

## Physical Properties

Property code	Value	Unit	Source
gf	21.35	kJ/mol	Joback Method
hf	-355.30	kJ/mol	Joback Method
hfus	32.30	kJ/mol	Joback Method
hvap	64.08	kJ/mol	Joback Method
log10ws	-4.63		Crippen Method
logp	4.302		Crippen Method
mvol	223.210	ml/mol	McGowan Method
pc	1783.35	kPa	Joback Method
rinpol	1903.00		NIST Webbook
rinpol	1903.00		NIST Webbook
tb	692.96	K	Joback Method
tc	900.69	K	Joback Method
tf	378.63	K	Joback Method
vc	0.854	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	642.91	J/molxK	692.96	Joback Method
cpg	725.11	J/molxK	866.06	Joback Method
cpg	710.76	J/molxK	831.44	Joback Method
cpg	695.43	J/molxK	796.82	Joback Method
cpg	679.04	J/molxK	762.20	Joback Method
cpg	661.56	J/molxK	727.58	Joback Method
cpg	738.52	J/molxK	900.69	Joback Method
dvisc	0.0003267	Paxs	692.96	Joback Method

dvisc	0.0003940	Paxs	640.57	Joback Method
dvisc	0.0004913	Paxs	588.18	Joback Method
dvisc	0.0006396	Paxs	535.80	Joback Method
dvisc	0.0008817	Paxs	483.41	Joback Method
dvisc	0.0013142	Paxs	431.02	Joback Method
dvisc	0.0021874	Paxs	378.63	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406840&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406840&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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