

# Diglycolic acid, isobutyl 2-isopropoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C17H24O6/c1-12(2)9-21-16(18)10-20-11-17(19)23-15-8-6-5-7-14(15)22-13(3)-
<b>InchiKey:</b>	BTIOTMSPINGVQU-UHFFFAOYSA-N
<b>Formula:</b>	C17H24O6
<b>SMILES:</b>	CC(C)COC(=O)COCC(=O)Oc1ccccc1OC(C)C
<b>Mol. weight [g/mol]:</b>	324.37

## Physical Properties

Property code	Value	Unit	Source
gf	-487.68	kJ/mol	Joback Method
hf	-933.75	kJ/mol	Joback Method
hfus	34.34	kJ/mol	Joback Method
hvap	78.73	kJ/mol	Joback Method
log10ws	-3.07		Crippen Method
logp	2.595		Crippen Method
mcvol	253.250	ml/mol	McGowan Method
pc	1652.46	kPa	Joback Method
rinpola	2789.00		NIST Webbook
rinpola	2789.00		NIST Webbook
tb	816.56	K	Joback Method
tc	1021.49	K	Joback Method
tf	479.07	K	Joback Method
vc	0.952	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	762.06	J/molxK	816.56	Joback Method
cpg	776.94	J/molxK	850.71	Joback Method
cpg	790.64	J/molxK	884.87	Joback Method
cpg	803.14	J/molxK	919.02	Joback Method
cpg	814.43	J/molxK	953.18	Joback Method
cpg	824.50	J/molxK	987.33	Joback Method
cpg	833.34	J/molxK	1021.49	Joback Method
dvisc	0.0005055	Paxs	479.07	Joback Method

dvisc	0.0002647	Paxs	535.32	Joback Method
dvisc	0.0001568	Paxs	591.57	Joback Method
dvisc	0.0001017	Paxs	647.82	Joback Method
dvisc	0.0000707	Paxs	704.06	Joback Method
dvisc	0.0000518	Paxs	760.31	Joback Method
dvisc	0.0000397	Paxs	816.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381980&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381980&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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