

# Diglycolic acid, 4-methoxybenzyl pentyl ester

<b>Inchi:</b>	InChI=1S/C17H24O6/c1-3-4-5-10-22-16(18)12-21-13-17(19)23-11-14-6-8-15(20-2)9-7-14
<b>InchiKey:</b>	CEUBMSDIGXBWES-UHFFFAOYSA-N
<b>Formula:</b>	C17H24O6
<b>SMILES:</b>	CCCCCOC(=O)COCC(=O)OCc1ccc(OC)cc1
<b>Mol. weight [g/mol]:</b>	324.37

## Physical Properties

Property code	Value	Unit	Source
gf	-482.80	kJ/mol	Joback Method
hf	-923.19	kJ/mol	Joback Method
hfus	41.39	kJ/mol	Joback Method
hvap	79.51	kJ/mol	Joback Method
log10ws	-3.05		Crippen Method
logp	2.488		Crippen Method
mvol	253.250	ml/mol	McGowan Method
pc	1631.17	kPa	Joback Method
rmpol	3065.00		NIST Webbook
rmpol	3065.00		NIST Webbook
tb	817.44	K	Joback Method
tc	1017.95	K	Joback Method
tf	509.07	K	Joback Method
vc	0.964	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	760.96	J/molxK	817.44	Joback Method
cpg	775.54	J/molxK	850.86	Joback Method
cpg	789.01	J/molxK	884.28	Joback Method
cpg	801.34	J/molxK	917.70	Joback Method
cpg	812.54	J/molxK	951.12	Joback Method
cpg	822.59	J/molxK	984.53	Joback Method
cpg	831.48	J/molxK	1017.95	Joback Method
dvisc	0.0003963	Paxs	509.07	Joback Method

dvisc	0.0002361	Paxs	560.47	Joback Method
dvisc	0.0001535	Paxs	611.86	Joback Method
dvisc	0.0001067	Paxs	663.25	Joback Method
dvisc	0.0000781	Paxs	714.65	Joback Method
dvisc	0.0000596	Paxs	766.04	Joback Method
dvisc	0.0000471	Paxs	817.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382249&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382249&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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