

# (S)-1,1-bis(4-Methoxyphenyl)propan-2-ol

<b>Inchi:</b>	InChI=1S/C17H20O3/c1-12(18)17(13-4-8-15(19-2)9-5-13)14-6-10-16(20-3)11-7-14/h4-12
<b>InchiKey:</b>	NHUSVJTYMHDUAP-LBPRGKRZSA-N
<b>Formula:</b>	C17H20O3
<b>SMILES:</b>	COc1ccc(C(c2ccc(OC)cc2)C(C)O)cc1
<b>Mol. weight [g/mol]:</b>	272.34
<b>CAS:</b>	212516-39-7

## Physical Properties

Property code	Value	Unit	Source
gf	-53.88	kJ/mol	Joback Method
hf	-371.32	kJ/mol	Joback Method
hfus	26.51	kJ/mol	Joback Method
hvap	80.04	kJ/mol	Joback Method
log10ws	-3.99		Crippen Method
logp	3.216		Crippen Method
mcvol	220.480	ml/mol	McGowan Method
pc	2175.46	kPa	Joback Method
rinpol	2276.50		NIST Webbook
rinpol	2276.50		NIST Webbook
tb	787.82	K	Joback Method
tc	1002.94	K	Joback Method
tf	434.51	K	Joback Method
vc	0.815	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	639.91	J/molxK	787.82	Joback Method
cpg	654.32	J/molxK	823.67	Joback Method
cpg	667.64	J/molxK	859.53	Joback Method
cpg	679.92	J/molxK	895.38	Joback Method
cpg	691.18	J/molxK	931.23	Joback Method
cpg	701.43	J/molxK	967.08	Joback Method
cpg	710.71	J/molxK	1002.94	Joback Method

dvisc	0.0007731	Paxs	434.51	Joback Method
dvisc	0.0002638	Paxs	493.39	Joback Method
dvisc	0.0001132	Paxs	552.28	Joback Method
dvisc	0.0000572	Paxs	611.16	Joback Method
dvisc	0.0000326	Paxs	670.05	Joback Method
dvisc	0.0000203	Paxs	728.93	Joback Method
dvisc	0.0000136	Paxs	787.82	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C212516397&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C212516397&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/90-482-7/S-1-1-bis-4-Methoxyphenyl-propan-2-ol.pdf>

Generated by Cheméo on 2025-12-05 19:15:42.787658681 +0000 UTC m=+4710340.317699335.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.