

(3S,3aS,6R,7R,9aS)-6-Hydroxy-1,1,7-trimethyldeca

Inchi:
acetate

InChI=1S/C17H28O3/c1-11(18)20-14-9-15(2,3)12-5-7-16(4)10-17(12,14)8-6-13(16)19/h1

InchiKey:

BAANKNZYZUSIFA-UHFFFAOYSA-N

Formula:

C17H28O3

SMILES:

CC(=O)OC1CC(C)(C)C2CCC3(C)CC12CCC3O

Mol. weight [g/mol]:

280.40

CAS:

127156-28-9

Physical Properties

Property code	Value	Unit	Source
gf	-172.13	kJ/mol	Joback Method
hf	-606.62	kJ/mol	Joback Method
hfus	19.09	kJ/mol	Joback Method
hvap	75.15	kJ/mol	Joback Method
log10ws	-4.01		Crippen Method
logp	3.296		Crippen Method
mcvol	231.120	ml/mol	McGowan Method
pc	2045.61	kPa	Joback Method
rinpol	1982.20		NIST Webbook
rinpol	1982.20		NIST Webbook
tb	776.57	K	Joback Method
tc	993.74	K	Joback Method
tf	516.57	K	Joback Method
vc	0.869	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	767.08	J/molxK	776.57	Joback Method
cpg	788.49	J/molxK	812.76	Joback Method
cpg	810.06	J/molxK	848.96	Joback Method
cpg	832.12	J/molxK	885.15	Joback Method
cpg	854.99	J/molxK	921.35	Joback Method
cpg	879.01	J/molxK	957.54	Joback Method
cpg	904.50	J/molxK	993.74	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C127156289&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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