

# Benzene, 1,3-dimethoxy-2-nonyl

<b>Inchi:</b>	InChI=1S/C17H28O2/c1-4-5-6-7-8-9-10-12-15-16(18-2)13-11-14-17(15)19-3/h11,13-14H
<b>InchiKey:</b>	VFVMVMNLTQHDCD-UHFFFAOYSA-N
<b>Formula:</b>	C17H28O2
<b>SMILES:</b>	CCCCCCCCC1c(OC)cccc1OC
<b>Mol. weight [g/mol]:</b>	264.40

## Physical Properties

Property code	Value	Unit	Source
gf	-24.59	kJ/mol	Joback Method
hf	-445.06	kJ/mol	Joback Method
hfus	35.42	kJ/mol	Joback Method
hvap	61.86	kJ/mol	Joback Method
log10ws	-5.44		Crippen Method
logp	4.997		Crippen Method
mcvol	238.370	ml/mol	McGowan Method
pc	1498.83	kPa	Joback Method
rinpol	1909.00		NIST Webbook
rinpol	1909.00		NIST Webbook
tb	669.84	K	Joback Method
tc	857.12	K	Joback Method
tf	377.27	K	Joback Method
vc	0.915	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	658.97	J/molxK	669.84	Joback Method
cpg	677.24	J/molxK	701.05	Joback Method
cpg	694.63	J/molxK	732.27	Joback Method
cpg	711.13	J/molxK	763.48	Joback Method
cpg	726.77	J/molxK	794.69	Joback Method
cpg	741.55	J/molxK	825.91	Joback Method
cpg	755.49	J/molxK	857.12	Joback Method
dvisc	0.0009250	Paxs	377.27	Joback Method

dvisc	0.0004912	Paxs	426.03	Joback Method
dvisc	0.0002971	Paxs	474.79	Joback Method
dvisc	0.0001973	Paxs	523.56	Joback Method
dvisc	0.0001405	Paxs	572.32	Joback Method
dvisc	0.0001055	Paxs	621.08	Joback Method
dvisc	0.0000826	Paxs	669.84	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R142917&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R142917&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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