

# Benzene, 1,3-dimethoxy-2-methyl-4-octyl

<b>Inchi:</b>	InChI=1S/C17H28O2/c1-5-6-7-8-9-10-11-15-12-13-16(18-3)14(2)17(15)19-4/h12-13H,5-
<b>InchiKey:</b>	XZNNMRJDQHZWGU-UHFFFAOYSA-N
<b>Formula:</b>	C17H28O2
<b>SMILES:</b>	CCCCCCCCc1ccc(OC)c(C)c1OC
<b>Mol. weight [g/mol]:</b>	264.40

## Physical Properties

Property code	Value	Unit	Source
gf	-34.22	kJ/mol	Joback Method
hf	-456.53	kJ/mol	Joback Method
hfus	35.04	kJ/mol	Joback Method
hvap	62.52	kJ/mol	Joback Method
log10ws	-5.50		Crippen Method
logp	4.915		Crippen Method
mvol	238.370	ml/mol	McGowan Method
pc	1481.57	kPa	Joback Method
rinpol	1912.00		NIST Webbook
rinpol	1912.00		NIST Webbook
tb	674.82	K	Joback Method
tc	862.96	K	Joback Method
tf	389.79	K	Joback Method
vc	0.915	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	658.98	J/molxK	674.82	Joback Method
cpg	677.12	J/molxK	706.18	Joback Method
cpg	694.38	J/molxK	737.53	Joback Method
cpg	710.79	J/molxK	768.89	Joback Method
cpg	726.34	J/molxK	800.24	Joback Method
cpg	741.04	J/molxK	831.60	Joback Method
cpg	754.90	J/molxK	862.96	Joback Method
dvisc	0.0007460	Paxs	389.79	Joback Method

dvisc	0.0004231	Paxs	437.30	Joback Method
dvisc	0.0002682	Paxs	484.80	Joback Method
dvisc	0.0001844	Paxs	532.30	Joback Method
dvisc	0.0001348	Paxs	579.81	Joback Method
dvisc	0.0001033	Paxs	627.31	Joback Method
dvisc	0.0000822	Paxs	674.82	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R142888&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R142888&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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