

# 4-Butylbenzoic acid, octadecyl ester

**Inchi:** InChI=1S/C29H50O2/c1-3-5-7-8-9-10-11-12-13-14-15-16-17-18-19-20-26-31-29(30)28-2  
**InchiKey:** PKAPDTSUVWQODK-UHFFFAOYSA-N  
**Formula:** C29H50O2  
**SMILES:** CCCCCCCCCCCCCCCCCOC(=O)c1ccc(CCCC)cc1  
**Mol. weight [g/mol]:** 430.71

## Physical Properties

Property code	Value	Unit	Source
gf	62.16	kJ/mol	Joback Method
hf	-661.63	kJ/mol	Joback Method
hfus	67.30	kJ/mol	Joback Method
hvap	92.24	kJ/mol	Joback Method
log10ws	-10.47		Crippen Method
logp	9.447		Crippen Method
mvol	403.150	ml/mol	McGowan Method
pc	754.74	kPa	Joback Method
rinpol	3232.90		NIST Webbook
rinpol	3232.90		NIST Webbook
tb	970.87	K	Joback Method
tc	1190.94	K	Joback Method
tf	527.69	K	Joback Method
vc	1.575	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1384.47	J/molxK	970.87	Joback Method
cpg	1475.81	J/molxK	1154.26	Joback Method
cpg	1460.29	J/molxK	1117.58	Joback Method
cpg	1443.48	J/molxK	1080.90	Joback Method
cpg	1425.29	J/molxK	1044.23	Joback Method
cpg	1405.65	J/molxK	1007.55	Joback Method
cpg	1490.12	J/molxK	1190.94	Joback Method
dvisc	0.0000208	Paxs	970.87	Joback Method

dvisc	0.0000277	Paxs	897.01	Joback Method
dvisc	0.0000389	Paxs	823.14	Joback Method
dvisc	0.0000584	Paxs	749.28	Joback Method
dvisc	0.0000958	Paxs	675.42	Joback Method
dvisc	0.0001775	Paxs	601.55	Joback Method
dvisc	0.0003908	Paxs	527.69	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292330&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292330&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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