

# 1,2-Cyclohexanedicarboxylic acid, 2-biphenyl nonyl ester

<b>Inchi:</b>	InChI=1S/C29H38O4/c1-2-3-4-5-6-7-15-22-32-28(30)25-19-11-12-20-26(25)29(31)33-27
<b>InchiKey:</b>	GVYZYXOTSDAQGP-UHFFFAOYSA-N
<b>Formula:</b>	C29H38O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)C1CCCCC1C(=O)Oc1ccccc1-c1ccccc1
<b>Mol. weight [g/mol]:</b>	450.61

## Physical Properties

Property code	Value	Unit	Source
gf	-42.61	kJ/mol	Joback Method
hf	-635.92	kJ/mol	Joback Method
hfus	57.04	kJ/mol	Joback Method
hvap	103.79	kJ/mol	Joback Method
log10ws	-8.94		Crippen Method
logp	7.359		Crippen Method
mcvol	375.970	ml/mol	McGowan Method
pc	1046.65	kPa	Joback Method
rinpol	3287.00		NIST Webbook
rinpol	3287.00		NIST Webbook
tb	1088.72	K	Joback Method
tc	1334.05	K	Joback Method
tf	629.41	K	Joback Method
vc	1.423	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1318.26	J/molxK	1088.72	Joback Method
cpg	1365.96	J/molxK	1293.16	Joback Method
cpg	1360.02	J/molxK	1252.27	Joback Method
cpg	1352.37	J/molxK	1211.38	Joback Method
cpg	1342.92	J/molxK	1170.50	Joback Method
cpg	1331.58	J/molxK	1129.61	Joback Method
cpg	1370.27	J/molxK	1334.05	Joback Method
dvisc	0.0000209	Paxs	1088.72	Joback Method

dvisc	0.0000270	Paxs	1012.17	Joback Method
dvisc	0.0000363	Paxs	935.62	Joback Method
dvisc	0.0000515	Paxs	859.07	Joback Method
dvisc	0.0000782	Paxs	782.51	Joback Method
dvisc	0.0001300	Paxs	705.96	Joback Method
dvisc	0.0002446	Paxs	629.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339600&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339600&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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