

# Benzenemethanol, 3-phenoxy-

<b>Other names:</b>	m-Phenoxybenzyl alcohol 3-Phenoxybenzenemethanol 3-Phenoxybenzyl alcohol Benzyl alcohol, m-phenoxy- 3-(Hydroxymethyl)diphenyl ether (3-Phenoxyphenyl)methanol 3-phenoxybenzylic alcohol
<b>Inchi:</b>	InChI=1S/C13H12O2/c14-10-11-5-4-8-13(9-11)15-12-6-2-1-3-7-12/h1-9,14H,10H2
<b>InchiKey:</b>	KGANAERDZBAECK-UHFFFAOYSA-N
<b>Formula:</b>	C13H12O2
<b>SMILES:</b>	OCc1cccc(Oc2ccccc2)c1
<b>Mol. weight [g/mol]:</b>	200.23
<b>CAS:</b>	13826-35-2

## Physical Properties

Property code	Value	Unit	Source
gf	31.95	kJ/mol	Joback Method
hf	-134.51	kJ/mol	Joback Method
hfus	22.39	kJ/mol	Joback Method
hvap	68.84	kJ/mol	Joback Method
log10ws	-3.25		Crippen Method
logp	2.971		Crippen Method
mvol	158.250	ml/mol	McGowan Method
pc	3287.81	kPa	Joback Method
rinpol	1836.50		NIST Webbook
rinpol	1836.50		NIST Webbook
tb	669.78	K	Joback Method
tc	891.53	K	Joback Method
tf	384.68	K	Joback Method
vc	0.585	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	402.07	J/molxK	669.78	Joback Method
cpg	457.27	J/molxK	854.57	Joback Method
cpg	447.91	J/molxK	817.61	Joback Method
cpg	437.74	J/molxK	780.66	Joback Method
cpg	426.74	J/molxK	743.70	Joback Method
cpg	414.86	J/molxK	706.74	Joback Method
cpg	465.88	J/molxK	891.53	Joback Method
dvisc	0.0000427	Paxs	669.78	Joback Method
dvisc	0.0000628	Paxs	622.26	Joback Method
dvisc	0.0000985	Paxs	574.75	Joback Method
dvisc	0.0001677	Paxs	527.23	Joback Method
dvisc	0.0003172	Paxs	479.71	Joback Method
dvisc	0.0006901	Paxs	432.20	Joback Method
dvisc	0.0018192	Paxs	384.68	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13826352&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13826352&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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