

1-Naphthalenecarboxaldehyde, 2-ethoxy-

Other names:	1-Naphthaldehyde, 2-ethoxy- 2-Ethoxy-1-naphthaldehyde 2-ethoxynaphthalene-1-carbaldehyde
Inchi:	InChI=1S/C13H12O2/c1-2-15-13-8-7-10-5-3-4-6-11(10)12(13)9-14/h3-9H,2H2,1H3
InchiKey:	IMNKQTWVJHODOS-UHFFFAOYSA-N
Formula:	C13H12O2
SMILES:	CCOc1ccc2ccccc2c1C=O
Mol. weight [g/mol]:	200.23
CAS:	19523-57-0

Physical Properties

Property code	Value	Unit	Source
gf	53.86	kJ/mol	Joback Method
hf	-124.79	kJ/mol	Joback Method
hfus	23.19	kJ/mol	Joback Method
hvap	58.90	kJ/mol	Joback Method
log10ws	-4.06		Crippen Method
logp	3.051		Crippen Method
mcvol	158.250	ml/mol	McGowan Method
pc	2881.21	kPa	Joback Method
tb	623.54	K	Joback Method
tc	849.94	K	Joback Method
tf	384.66	K	Joback Method
vc	0.613	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	384.44	J/molxK	623.54	Joback Method
cpg	397.88	J/molxK	661.27	Joback Method
cpg	410.42	J/molxK	699.01	Joback Method
cpg	422.12	J/molxK	736.74	Joback Method
cpg	433.01	J/molxK	774.47	Joback Method
cpg	443.14	J/molxK	812.20	Joback Method

cpg	452.56	J/mol×K	849.94	Joback Method
dvisc	0.0013212	Paxs	384.66	Joback Method
dvisc	0.0009204	Paxs	424.47	Joback Method
dvisc	0.0006822	Paxs	464.29	Joback Method
dvisc	0.0005301	Paxs	504.10	Joback Method
dvisc	0.0004274	Paxs	543.91	Joback Method
dvisc	0.0003549	Paxs	583.73	Joback Method
dvisc	0.0003018	Paxs	623.54	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C19523570&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/90-653-7/1-Naphthalenecarboxaldehyde-2-ethoxy.pdf>

Generated by Cheméo on 2024-04-27 07:42:25.012131941 +0000 UTC m=+16492993.932709257.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.