

# Phenol, o-(o-methoxyphenoxy)-

<b>Other names:</b>	2-(2-Methoxyphenoxy)phenol 2-Hydroxy-2'-methoxy diphenyl ether
<b>Inchi:</b>	InChI=1S/C13H12O3/c1-15-12-8-4-5-9-13(12)16-11-7-3-2-6-10(11)14/h2-9,14H,1H3
<b>InchiKey:</b>	OHMCFTVLHBWELH-UHFFFAOYSA-N
<b>Formula:</b>	C13H12O3
<b>SMILES:</b>	COc1ccccc1Oc1ccccc1O
<b>Mol. weight [g/mol]:</b>	216.23
<b>CAS:</b>	21905-60-2

## Physical Properties

Property code	Value	Unit	Source
gf	-90.85	kJ/mol	Joback Method
hf	-291.81	kJ/mol	Joback Method
hfus	25.28	kJ/mol	Joback Method
hvap	67.58	kJ/mol	Joback Method
log10ws	-2.77		Crippen Method
logp	3.193		Crippen Method
mvol	164.120	ml/mol	McGowan Method
pc	3443.98	kPa	Joback Method
tb	680.64	K	Joback Method
tc	927.48	K	Joback Method
tf	457.81	K	Joback Method
vc	0.549	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	424.64	J/molxK	680.64	Joback Method
cpg	438.41	J/molxK	721.78	Joback Method
cpg	451.17	J/molxK	762.92	Joback Method
cpg	463.02	J/molxK	804.06	Joback Method
cpg	474.03	J/molxK	845.20	Joback Method
cpg	484.28	J/molxK	886.34	Joback Method
cpg	493.86	J/molxK	927.48	Joback Method

dvisc	0.0002772	Paxs	457.81	Joback Method
dvisc	0.0001369	Paxs	494.95	Joback Method
dvisc	0.0000746	Paxs	532.09	Joback Method
dvisc	0.0000440	Paxs	569.23	Joback Method
dvisc	0.0000277	Paxs	606.36	Joback Method
dvisc	0.0000184	Paxs	643.50	Joback Method
dvisc	0.0000127	Paxs	680.64	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21905602&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21905602&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/90-655-5/Phenol-o-o-methoxyphenoxy.pdf>

Generated by Cheméo on 2024-04-19 13:52:23.724392182 +0000 UTC m=+15823992.644969494.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.