

# Benzamide, 3-methyl-N-pentyl-

<b>Inchi:</b>	InChI=1S/C13H19NO/c1-3-4-5-9-14-13(15)12-8-6-7-11(2)10-12/h6-8,10H,3-5,9H2,1-2H3
<b>InchiKey:</b>	DQFOMLXSBIEIBP-UHFFFAOYSA-N
<b>Formula:</b>	C13H19NO
<b>SMILES:</b>	CCCCCN=C(O)c1cccc(C)c1
<b>Mol. weight [g/mol]:</b>	205.30

## Physical Properties

Property code	Value	Unit	Source
hf	-166.39	kJ/mol	Joback Method
hvap	67.54	kJ/mol	Joback Method
log10ws	-3.51		Crippen Method
logp	3.490		Crippen Method
mcvol	181.820	ml/mol	McGowan Method
pc	2147.32	kPa	Joback Method
rinpol	1869.00		NIST Webbook
rinpol	1869.00		NIST Webbook
tb	697.24	K	Joback Method
tc	900.40	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U407413&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U407413&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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