

# Phenylacetamide, N-pentyl-

<b>Inchi:</b>	InChI=1S/C13H19NO/c1-2-3-7-10-14-13(15)11-12-8-5-4-6-9-12/h4-6,8-9H,2-3,7,10-11H2
<b>InchiKey:</b>	YZLZQXXVYJDVEZ-UHFFFAOYSA-N
<b>Formula:</b>	C13H19NO
<b>SMILES:</b>	CCCCCN=C(O)Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	205.30

## Physical Properties

Property code	Value	Unit	Source
hf	-154.92	kJ/mol	Joback Method
hvap	66.88	kJ/mol	Joback Method
log10ws	-3.35		Crippen Method
logp	3.376		Crippen Method
mcvol	181.820	ml/mol	McGowan Method
pc	2177.49	kPa	Joback Method
rinpol	1791.00		NIST Webbook
rinpol	1791.00		NIST Webbook
tb	692.26	K	Joback Method
tc	894.59	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U407224&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U407224&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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