

# 2-Cyclohexen-1-one, 3,5,5-trimethyl-4-(3-oxo-1-butenyl)-

**Other names:** 4-(3-oxo-1-butenyl)-3,5,5-trimethyl-2-cyclohexen-1-one  
3,5,5-trimethyl-4-(3-oxo-1-butenyl)-2-cyclohexen-1-one  
3,5,5-Trimethyl-4-(3-oxobut-1-en-1-yl)cyclohex-2-enone

**Inchi:** InChI=1S/C13H18O2/c1-9-7-11(15)8-13(3,4)12(9)6-5-10(2)14/h5-7,12H,8H2,1-4H3/b6-5

**InchiKey:** MLYOGKJJENFVJN-AATRIKPKSA-N

**Formula:** C13H18O2

**SMILES:** CC(=O)C=CC1C(C)=CC(=O)CC1(C)C

**Mol. weight [g/mol]:** 206.28

**CAS:** 20194-68-7

## Physical Properties

Property code	Value	Unit	Source
gf	-81.13	kJ/mol	Joback Method
hf	-349.18	kJ/mol	Joback Method
hfus	18.18	kJ/mol	Joback Method
hvap	55.41	kJ/mol	Joback Method
log10ws	-2.94		Crippen Method
logp	2.693		Crippen Method
mcvol	177.710	ml/mol	McGowan Method
pc	2315.84	kPa	Joback Method
rinpol	1661.10		NIST Webbook
rinpol	1661.10		NIST Webbook
ripol	2067.00		NIST Webbook
ripol	2078.00		NIST Webbook
ripol	2067.00		NIST Webbook
ripol	2078.00		NIST Webbook
tb	641.95	K	Joback Method
tc	872.71	K	Joback Method
tf	389.66	K	Joback Method
vc	0.672	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	471.42	J/mol×K	641.95	Joback Method
cpg	489.28	J/mol×K	680.41	Joback Method
cpg	506.22	J/mol×K	718.87	Joback Method
cpg	522.33	J/mol×K	757.33	Joback Method
cpg	537.74	J/mol×K	795.79	Joback Method
cpg	552.56	J/mol×K	834.25	Joback Method
cpg	566.89	J/mol×K	872.71	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20194687&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20194687&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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