

M-xylene-alpha,alpha'-diol, 2-hydroxy-5-(3-methyl-2-buten-1-yl)-

Inchi:	InChI=1S/C13H18O3/c1-9(2)3-4-10-5-11(7-14)13(16)12(6-10)8-15/h3,5-6,14-16H,4,7-8H
InchiKey:	NCWXUUYNQITKOB-UHFFFAOYSA-N
Formula:	C13H18O3
SMILES:	CC(C)=CCc1cc(CO)c(O)c(CO)c1
Mol. weight [g/mol]:	222.28
CAS:	116632-93-0

Physical Properties

Property code	Value	Unit	Source
gf	-204.86	kJ/mol	Joback Method
hf	-472.40	kJ/mol	Joback Method
hfus	35.54	kJ/mol	Joback Method
hvap	94.54	kJ/mol	Joback Method
log10ws	-3.22		Crippen Method
logp	1.885		Crippen Method
mcvol	183.580	ml/mol	McGowan Method
pc	3173.97	kPa	Joback Method
tb	802.50	K	Joback Method
tc	1002.34	K	Joback Method
tf	502.05	K	Joback Method
vc	0.640	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	535.90	J/molxK	802.50	Joback Method
cpg	546.50	J/molxK	835.81	Joback Method
cpg	556.70	J/molxK	869.11	Joback Method
cpg	566.58	J/molxK	902.42	Joback Method
cpg	576.20	J/molxK	935.73	Joback Method
cpg	585.64	J/molxK	969.03	Joback Method
cpg	594.97	J/molxK	1002.34	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C116632930&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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