

# 6-(4-Methoxyphenyl)hex-3-en-2-one

<b>Inchi:</b>	InChI=1S/C13H16O2/c1-11(14)5-3-4-6-12-7-9-13(15-2)10-8-12/h3,5,7-10H,4,6H2,1-2H3
<b>InchiKey:</b>	VHCHKVITYZDWNGL-HWKANZROSA-N
<b>Formula:</b>	C13H16O2
<b>SMILES:</b>	COc1ccc(CCC=CC(C)=O)cc1
<b>Mol. weight [g/mol]:</b>	204.26
<b>CAS:</b>	118790-84-4

## Physical Properties

Property code	Value	Unit	Source
gf	7.66	kJ/mol	Joback Method
hf	-214.17	kJ/mol	Joback Method
hfus	26.07	kJ/mol	Joback Method
hvap	56.58	kJ/mol	Joback Method
log10ws	-3.20		Crippen Method
logp	2.773		Crippen Method
mcvol	173.410	ml/mol	McGowan Method
pc	2374.90	kPa	Joback Method
rinpol	1755.20		NIST Webbook
rinpol	1755.20		NIST Webbook
tb	608.95	K	Joback Method
tc	821.29	K	Joback Method
tf	342.29	K	Joback Method
vc	0.659	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	422.62	J/molxK	608.95	Joback Method
cpg	437.68	J/molxK	644.34	Joback Method
cpg	451.85	J/molxK	679.73	Joback Method
cpg	465.17	J/molxK	715.12	Joback Method
cpg	477.67	J/molxK	750.51	Joback Method
cpg	489.38	J/molxK	785.90	Joback Method
cpg	500.34	J/molxK	821.29	Joback Method

dvisc	0.0015261	Paxs	342.29	Joback Method
dvisc	0.0008169	Paxs	386.73	Joback Method
dvisc	0.0004975	Paxs	431.18	Joback Method
dvisc	0.0003323	Paxs	475.62	Joback Method
dvisc	0.0002379	Paxs	520.06	Joback Method
dvisc	0.0001795	Paxs	564.51	Joback Method
dvisc	0.0001411	Paxs	608.95	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C118790844&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C118790844&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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