

# Diglycolic acid, ethyl 2-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C13H16O6/c1-3-18-12(14)8-17-9-13(15)19-11-7-5-4-6-10(11)16-2/h4-7H,3,8-9
<b>InchiKey:</b>	GCMFAVPDQOBISS-UHFFFAOYSA-N
<b>Formula:</b>	C13H16O6
<b>SMILES:</b>	CCOC(=O)COCC(=O)Oc1ccccc1OC
<b>Mol. weight [g/mol]:</b>	268.26

## Physical Properties

Property code	Value	Unit	Source
gf	-516.48	kJ/mol	Joback Method
hf	-840.63	kJ/mol	Joback Method
hfus	31.03	kJ/mol	Joback Method
hvap	70.60	kJ/mol	Joback Method
log10ws	-1.53		Crippen Method
logp	1.180		Crippen Method
mvol	196.890	ml/mol	McGowan Method
pc	2284.95	kPa	Joback Method
rinpol	2414.00		NIST Webbook
rinpol	2414.00		NIST Webbook
tb	725.92	K	Joback Method
tc	931.20	K	Joback Method
tf	463.99	K	Joback Method
vc	0.740	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	539.65	J/molxK	725.92	Joback Method
cpg	553.00	J/molxK	760.13	Joback Method
cpg	565.47	J/molxK	794.35	Joback Method
cpg	577.03	J/molxK	828.56	Joback Method
cpg	587.66	J/molxK	862.77	Joback Method
cpg	597.33	J/molxK	896.98	Joback Method
cpg	606.03	J/molxK	931.20	Joback Method
dvisc	0.0005583	Paxs	463.99	Joback Method

dvisc	0.0003515	Paxs	507.65	Joback Method
dvisc	0.0002381	Paxs	551.30	Joback Method
dvisc	0.0001707	Paxs	594.96	Joback Method
dvisc	0.0001281	Paxs	638.61	Joback Method
dvisc	0.0000998	Paxs	682.27	Joback Method
dvisc	0.0000801	Paxs	725.92	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382239&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382239&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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